

# Review on Thermocline Storage Effectiveness for Concentrating Solar Power Plant

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## Abstract

In this paper, a literature review on thermocline storage performance for Concentrating Solar Power (CSP) plant storage systems has been conducted. The efficiency of materials to store heat depends on the storage process like sensible heat storage, latent heat storage and thermochemical one and also on their properties. This study has been focused on sensible heat storage materials especially thermocline storage system (DMT) using eco-materials which has a high potentiality (35%) to reduce CSP cost. There is a possibility to use natural rocks, industry waste and to develop also materials for a thermocline storage within a bed called packed bed using one tank. The thermal storage materials should have some optimum parameters (particle diameter less than 2 cm and good thermo-physical properties) to achieve better thermal storage performance (thermal cycle efficiency, extraction factor). However, the size and the shape of natural rocks are uncontrollable (big diameter) and can drive to thermocline degradation, catastrophic thermal ratcheting and poor thermal stratification due to the variability of the storage system porosity and the stress on the storage tank wall. Also a better thermal storage efficiency is achievable at low velocity and with good thermo-physical properties of the HTF. The ratio H/D, the height, the porosity, the shape and the position of the tank should be optimized to increase the storage efficiency.

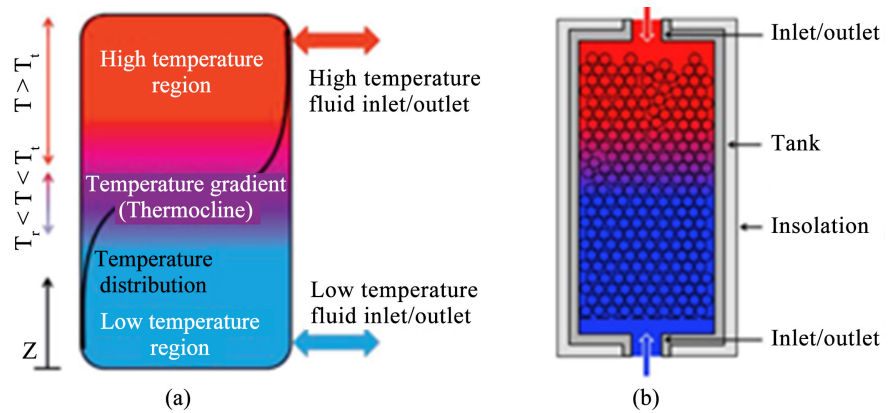
## Keywords

Thermocline Storage, Storage Effectiveness, Storage Materials, Optimum Parameters

## 1. Introduction

The global demand for energy is growing and conventional resources like coal and petroleum are depleting, renewable resources is expected to play a crucial role in the future. One promising of these renewable resources technologies is concentrating solar power (CSP) plants [1]. However, CSP technology affordability is suffering from energy storage system for electricity dispatching during the night-time and cloud cover. This is explained by the important initial investment of 15% to 20% due to the two-tank storage system and the high price of heat transfer fluid (HTF) like mineral oil and synthetic oil, use also as storage medium since 1984 at SEGS plant [2]. Another challenge of mineral oil is the high risk of flammability, for example in 2009 the solar field of Andasol CSP plant in Spain took fire [3]. To face these challenges in 2004, an engineering study has shown the possibility of reduction of 10% of levelized electricity cost using molten salt in the medium storage [2]. Become the most effective thermal energy storage material, most CSP plant like Andasol, Ouarzazate have molten salt as storage medium material [4] [5]. The molten salt is also compromised because of its high melting temperature and the complex management of its thermal stability [3] [6] [7]. Apart from the high cost, low thermal stability of mineral oil and molten salt, these materials have a negative impact on the environment. Though, studies have shown that there is a possibility to have a broad reduction of CSP cost (33% to 35%) by cancelling one tank of the storage system and using eco-materials (natural rocks, industrial waste) as filler materials [4] [8]. But about 33% of molten salt withdrawn from the thermocline are below the nominal temperature of storing [4]. This work aims to assess the effectiveness of thermocline storage system. To achieve this objective, we will assess the different thermocline storage materials. Look in the literature the output of different thermoclines projects, their efficiencies and the different parameters that influence thermocline efficiency to see how to improve its performances for CSP plants. A thermocline tank is one that uses a single tank to store thermal energy. There is thermocline tank in experimentation at Solar One, University of Bayreuth, Ait Baba, JÜLICH etc. [9] [10] [11] [12]. In this tank a thermal gradient separates the hot zone from the cold one. The word thermocline is from oceanic language. It designates the thermal transition zone between the superficial waters and the deep waters. At the top and the bottom of thermocline zone, temperatures are quasi-uniform, while the difference of temperature is important in the width of thermocline [3]. There is two type of thermocline system. When the tank use only fluid like molten salt, it is called a single-medium thermocline (SMT). To reduce the amount of molten salt required, a granulated material is added to have a dual-media thermocline (DMT) with economical and technical advantages over SMT tanks [13] as presented in **Figure 1**.

The single-medium thermocline is active storage system while the DMT can be both active and passive also called regenerators [12] [14]. When the thermal capacity of the heat transfer fluid (HTF) is significant and contribute to the



**Figure 1.** (a) Single-medium thermocline (SMT); (b) Dual-medium thermocline (DMT) [3] [4].

whole storage capacity, the DMT is a dual-medium system (active passive) [14] [15]. When gas or air is used as HTF the storage is passive.

In the following section the effectiveness on the different type of thermocline storage system are presented. With more focused on DMT system the effects of some key design parameters such as the inlet temperature and fluid flow (velocity), fluid properties, filler materials diameter and their thermal properties, tank porosity, and the tank aspect ratio, shape and position on the thermal performance of the thermocline TES are presented based on literature analysis.

## 2. Thermocline Systems and Their Effectiveness

### 2.1. Single-Medium Thermocline Systems and Their Effectiveness

The performance of a SMT is determine by the thermal stability, thermal capacity the maximal temperature of the HTF the melting point, compatibility of the storage material with the tank wall avoiding any corrosion [7] [16]. In case of indirect active storage, the HTF like oil should have a low vapour pressure, high flammability temperature, low cost and low impact on environment. One of the first CSP pilot plants has used one tank as thermocline storage system is Solar One since 1983. The strategy of this storage system was to supply auxiliary steam needs of the pilot power plant and experimentation for the thermal storage system. Oil has been used in the storage medium and heat transfer fluid through heat-exchanger. But in august 1986, this routine operation was interrupted because of the produced fire at the top of the tank due to over pressurization and the rupture of the thermal storage tank caused by significant quantities of water in the heat transfer fluid (oil) [17]. So most of STM systems are commonly used for industrial refrigeration systems and domestic solar thermal devices. To maintain under static and dynamic conditions a thermocline Abdouly *et al.* studying in 1982 the static and dynamic properties of a water thermocline carried out a possibility degradation of the thermocline due to heat mechanism losses through the insulation. Also, the tank should have an optimal size (length > 10 ft, diameter > 1.5 ft and insulation resistance > 20 hr·ft<sup>2</sup> F/B.t.u) [18]. Water has an excel-

lent specific heat (4.19 kJ/kg K) but the system need to be pressurized at above 100°C, which adds tremendously to cost, increase accident risk [19]. Exploration through numerical simulation has shown a higher thermal performance of molten salt SMT than the DMT one because there is absence of thermal ratcheting. But molten salt SMT tank is more sensitive to flow disturbances and have less cost-effectiveness [13]. The main equations governing take into account the motion of the fluid inside the tank that obeys the mass and momentum conservation principles as expressed in the following laminar flow equations. Darcy's and Forchheimer's terms must be included to account for viscous and momentum dissipation in the porous bed:

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho u) = 0 \quad (1)$$

$$\frac{\partial (\rho u)}{\partial t} + \nabla \cdot (uu) = -\Delta p + \nabla \tau + \rho g \quad (2)$$

One equation is required for a SMT tank. The effective thermal conductivity is equal to the thermal conductivity

$$\frac{\partial (\rho c_p T)}{\partial t} + \nabla \cdot (\rho u c_p T) = \nabla (\kappa \nabla T) + \mu \vartheta \quad (3)$$

where  $\rho$  is the density,  $c_p$  the specific heat,  $T$  fluid temperature,  $K$  thermal conductivity,  $\mu$  the viscosity of the fluid and  $\vartheta$  viscous dissipation function;  $\nabla$  defines the dimension of the model (1D, 2D or 3D).

Based on these models, researches have carried out the influence of the ratio height over diameter (H/D) and tank position on thermal stratification. In 2017 Burak *et al.* have found optimal values of 0.5° and 45° respectively for the ratio and tank position angle for water thermocline storage tank with 93% of stratification efficiency [20]. Bent inlet geometry design is required to have a good stratification for the Reynolds number between 7500 and 20,500 and for HTF velocity between 0.11 - 0.16 m/s [21].

## 2.2. Dual-Medium Thermocline System and Its Effectiveness

Most of CSP pilot and laboratory scale projects for thermocline storage studied are dual-medium thermocline [10] [11] [12] [17] [22] [23] [24]. DMT tanks are favored because they have economical and technical advantage. The filler material reduces the quantity of the liquid material (oil, molten salt) which is often the most expensive. But many parameters need to be considered to have a good thermal stratification therefore good extraction of the thermocline during charging and discharging time. Solar One CSP pilot plant is one of the first plant where DMT system has been experimented since 1982. The storage system was sized to generate 7 MWe during four hours. The thermal storage system employs dual liquid (Caloria HT-43) and solid (quartzite and sand) storage media in a thermal storage tank with 13.8 m of height, 18.3 m of diameter and 0.304 m of external insulation. The HTF (Caloria HT-43) was heat from 218°C to 304°C and was

heated indirectly through heat exchanger by the steam from the receiver. Thermal-to-electric efficiency was 25% from thermal storage steam instead of 35% because of the limitation to use oil more than 304°C (avoid vaporization) and the indirect storage method [25]. In 2012 the thermo mechanical performance simulation (infinite rigidity approach for the solid filler) of the Solar One thermocline storage tank by Scott linked the stresses along the tank wall to some finite amount of solid filler volume change due to thermal expansion and mechanical interactions; because the measured horizontal stress achieved 190 MPa against 177 MPa from numerical investigation (6.8% of deviation) [9]. Moreover the thermal performance of the molten salt packed-bed investigated in 2013 [26] carried out the impact of solid particles size on thermocline performance. The efficiency is the highest for the particles diameter smaller than 1.9 cm. The discharge time increase with thermal capacity (energy density) through different solid particle material. And also the particle should have an optimal thermal conductivity [27] to easier release heat during discharge time [26]. Solid particle like cast iron and ceramics have long discharge but may have issue of availability and cost effectiveness, while rock like quartzite rock have a good discharge efficiency and available with effective cost. That explains the use of quartzite/sand in the first thermocline storage tank (solar one) and why rocks are being experimented as thermal energy storage materials. But the uncontrolled shape and size impact the thermocline efficiency (big size, not well shape) [3]. In addition, the storage tank impacts the storage efficiency through insulation, its size and wall. The heat losses have to be taken into account for small tanks to correctly predict outlet temperature, but can be negligible for large-scale tank, also the tank wall may exert non-negligible effect on temperature profile inside the tank [28]. The thermal storage efficiency increase with the ratio height over diameter (H/D) and should have a minimum value of 2.3948 [29] because resulting in higher velocities of the HTF. And drive to a better heat transfer with the solid particles [27]. The optimal height of tank could be 30 m according to Angelini *et al.*, [4]. Also, different HTFs may influence the thermal performance of the thermocline through the porosity of the tank. When the porosity increase from 0.1 to 0.7 the effectiveness of the storage tank decrease and could be more suitable with thermol oil than molten salt like solar salt [30].

### 2.2.1. Governing Equation

To evaluate the effect of tank properties (size, shape, porosity, position), filler material properties and HTF properties on the performance or the behaviors of packed-bed thermocline system; most of the models developed in the literature are based on Schuman equations developed since 1929 [31]. This one-dimension model was used for porous medium modelling and combined two energy equations. One for the fluid and one for the solid particles.

$$\varepsilon \rho_f c_{pf} \left( \frac{\partial T_f}{\partial t} + u \frac{\partial T_f}{\partial z} \right) = h \cdot a_s \cdot (T_s - T_f) \quad (4)$$

$$(1-\varepsilon)\rho_s c_{ps} \cdot \frac{\partial T_s}{\partial t} = h \cdot a_s \cdot (T_f - T_s) \quad (5)$$

where  $a_s$  is representing solid surface per volume unit of the bed in  $\text{m}^2 \cdot \text{m}^{-3}$ ,  $T_s$  solid temperature,  $T_f$  and fluid temperature in  $^\circ\text{C}$ ,  $h$  convective heat capacity coefficient in  $\text{W} \cdot \text{m}^2 \cdot \text{K}^{-1}$ ,  $u$  the velocity of the fluid in  $\text{m/s}$ ,  $\varepsilon$  porosity of the porous medium,  $\rho_f$  the density of the fluid in  $\text{kg} \cdot \text{m}^{-3}$ ,  $c_{pf}$  the specific heat capacity of the fluid in  $\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$ .

$\rho_s$  the density of the solid in  $\text{kg} \cdot \text{m}^{-3}$ ,  $c_{ps}$  the specific heat capacity of the solid in  $\text{J} \cdot \text{kg}^{-1} \cdot \text{K}^{-1}$ .

For better understanding thermocline behaviors, the thermal losses, the effective conductivities of the fluid and solid, and tank thermal capacity have been added to Schuman model [24] [32]. And energy balance for the tank wall has been taken into account. Through the years to better analyze temperature distribution based on continuous solid phase assumption and fluid flow the model two dimension [33] and three dimension [9] [23] [34] [35] respectively have been developed. The general description of these models with three equations is presented as follow:

- Energy equation for the fluid

$$\varepsilon \rho_f c_{pf} \left( \frac{\partial T_f}{\partial t} + \nabla u \frac{\partial T_f}{\partial z} \right) = \nabla (\kappa_{f,eff} \nabla T_f) + h_v (T_s - T_f) + U_{u,fw} (T_w - T_f) \quad (6)$$

- Energy equation for the solid particles

$$(1-\varepsilon)\rho_s c_{ps} \cdot \frac{\partial T_s}{\partial t} = \nabla (\kappa_{s,eff} \nabla T_s) + h_{vs} (T_f - T_s) + U_{u,sw} (T_w - T_s) \quad (7)$$

- Energy equation for the tank wall.

$$\rho_w c_{pw} \cdot \frac{\partial T_w}{\partial t} = \nabla (\kappa_{w,eff} \nabla T_w) + U_{v,fw} (T_f - T_w) + U_{w,sw} (T_s - T_w) + h_{v,ext} (T_w - T_{ext}) \quad (8)$$

where  $h_v$  convective heat transfer coefficient per unit bed volume in  $\text{W} \cdot \text{m}^{-3} \cdot \text{K}^{-1}$ ,  $U_{u,fw}$  overall heat transfer coefficient between the tank wall and the fluid,  $T_w$  the tank wall temperature,  $U_{u,sw}$  overall heat transfer coefficient between the tank wall and the solid,  $h_{v,ext}$  convective heat transfer coefficient per unit bed volume in  $\text{W} \cdot \text{m}^{-3} \cdot \text{K}^{-1}$  with ambient air.

A part from energy equations, continuity equation and momentum equation should be added [26]. According to the model dimension and the size of the tank several numerical methods like finite volume, finite element and different finite have been used to resolve these energy balance equations and study the performance of a thermocline storage system during thermal cycles. The important of numerical model is to have data from an ideal system and use it for experimental data validation. The three equation models provide an effective thermal performance study of a storage system than one equation [24]. Two equations model can effectively study thermocline behaviors [36] like three equations model by reducing 18% of computation time [32].

### 2.2.2. Thermal Stratification

Stratification in the storage tank results from buoyancy effects cause by density differences in the storage. The hotter fluid rises to the top and the colder fluid falls to the bottom [37]. This thermal stratification is among the key factors that can sensibly affect the performance of the TES system: the thinner of the thermocline, the thermodynamic quality of the energy stored. The thermal stratification efficiency into packed-bed can be compute as follow:

$$\eta_{MIX} = \frac{\eta_{str} - \eta_{exp}}{\eta_{str} - \eta_{mix}} \quad (9)$$

where  $M_{exp}$  is the moment of the energy real TES,  $M_{str}$  the moment of the ideally stratified TES and  $M_{mix}$  fully mixed TES.

Into a packed bed as thermocline storage system using gravel as filler material and air as HTF, the thermal stratification efficiency enhances with starting-up of thermal cycles. Based on moment of energy and one equation model for energy balance resolved with finite volume approach and 3D CFD simulations software, Simone *et al.* [38] have shown a stable thermal stratification after 30 cycles with 93% as efficiency as showed in **Figure 2** and **Figure 3**. This shown the important of pre-charging the TES system before the first cycles to reduce the long time required to achieve a stable thermal stratification [39]. Similar results were achieved by Ortega *et al.* using air as HTF and rocks as filler materials [39].

### 2.2.3. Thermal Cycle Effectiveness

From **Figure 3** we can assume that the stratification efficiency could depend on the charge and discharge efficiency which is influenced by porosity of the tank, particles size, and storage capacity [27]. Moreover thermocline thickness defined as the covering length of the thermocline region depend on charging and discharging cycle and can be defined according to Chang *et al.* [40] by

$$dX = \min \{ H(T_{out}, T_{crit,h}) \} - H(T_{crit,C}) \text{ in dicharging cycle} \quad (10)$$

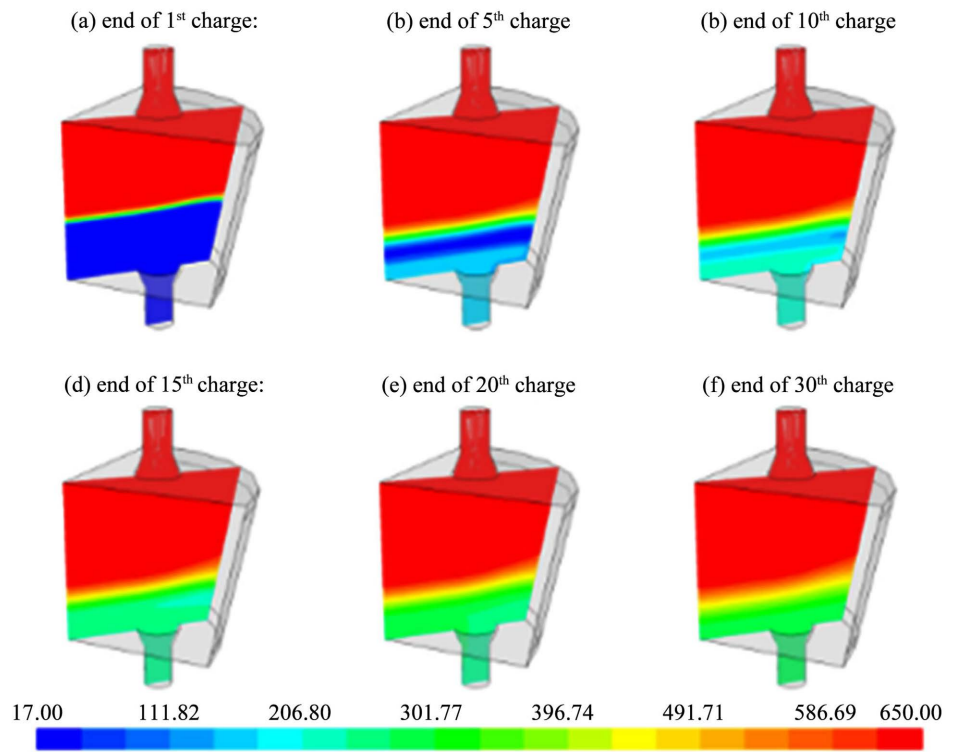
$$dX = H(T_{crit,C}) - \max \{ H(T_{out}, T_{crit,h}) \} \text{ in charging cycle} \quad (11)$$

where  $T_{crit,h} = T_h - 5$  and  $T_{crit,c} = T_c + 5$  Represent the critical low and hot temperature for evaluating the thermocline thickness respectively. From there we can realize that a thin thermocline can be achieved with an effective thermal stratification and thermal cycle. A thin thermocline thickness can conduct to a good cycle efficiency defined as follow:

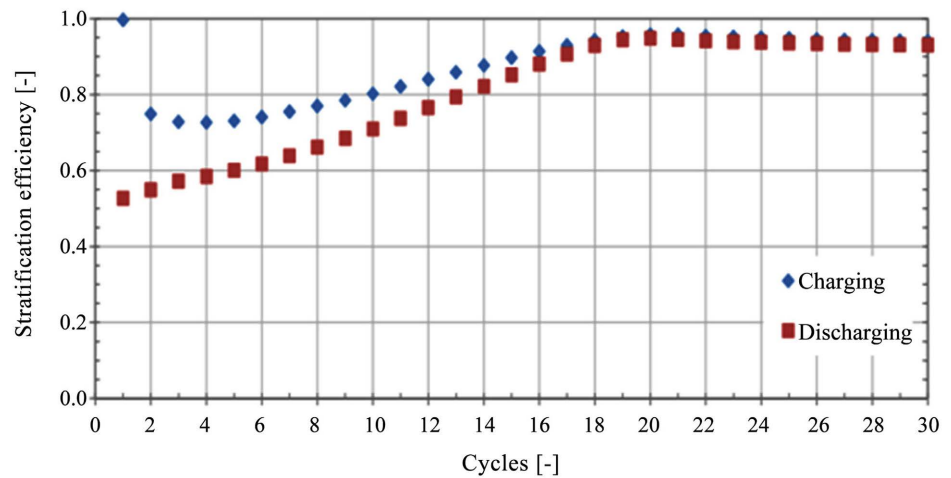
$$\eta = \frac{Q_{dish}}{Q_{chg}} = \frac{\int_0^{t_{dish}} \dot{m} c_{p,f} (T_{out} - T_{min}) dt}{\int_0^{t_{dish}} \dot{m} c_{p,f} (T_{max} - T_{min}) dt} \quad (12)$$

where  $\eta$  is the effective discharge efficiency,  $t_{dis} = t_{dis,ref}$  is the time at which the HTF temperature drops to a threshold value. The threshold value is usually determined by the application of the interest and is arbitrary chosen like  $T_h - 5$ ,  $T_h - 20$  etc.

An effective thermal storage depends on the discharge efficiency and the discharge time.



**Figure 2.** Thermal stratification enhancement with thermal cycles (end of each charge) [38].



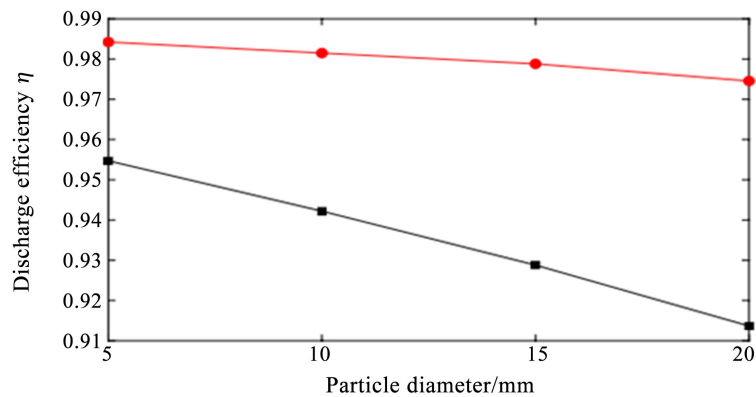
**Figure 3.** Average transient stratification efficiency for consecutive charge/discharge cycles [38].

### 3. Parameters Affecting Thermocline Storage Performance

#### 3.1. Particles Effect on Discharge Effectiveness

Based on transient two-dimensional dispersion-concentric (D-C) model for heat transfer and fluid dynamics in a packed-bed molten salt thermocline thermal storage system xu *et al.* [26] have shown the possibility reduction discharge efficiency [26] with large particles confirmed by numerous other authors [26] [27] [41] as showed in **Figure 4.**





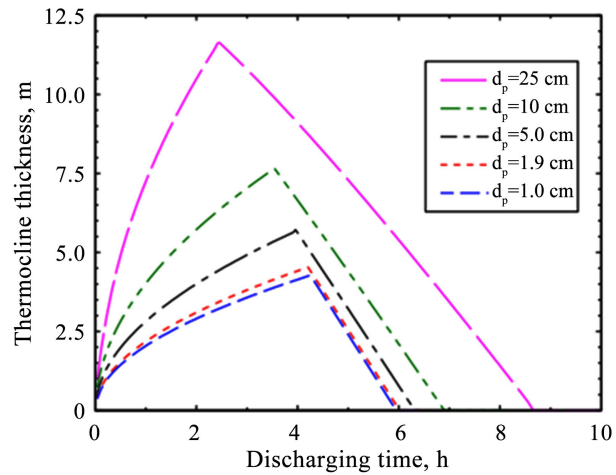
**Figure 4.** Effect of particle size on discharge efficiency [27].

To have a thin thermocline with an effective discharge efficiency and good discharge time, small particle diameter is required (less than 2 cm) to avoid bad thermal distribution in the particle and provide a good heat transfer process between the fluid and the particle, but when the particle diameter is smaller enough, the influence of particle diameter on thermal performance becomes negligible [26]. **Figure 5** presents the effect of particle diameter on thermocline efficiency.

A part from that, the thermal performance of a thermocline is influencing by the properties of the particles. The main properties governing the thermal performance of a filler material are the thermal conductivity, the density and volumetric heat capacity. Calcite and quartz are the main component governing these thermal and mechanical properties for natural rocks [42]. The thermal conductivity of the filler material influences the thermal distribution inside the tank. Greater thermal diffusion (increasing thermal lost) in DMT system may be explain by high thermal conductivity of the solid particle driving to bad thermal distribution and low thermal cycle efficiency [13]. This hypothesis can explain better the different results achieved by Keilany et al in 2018 using cofalit rock and aluminium sphere as filler materials than based on particles shape; the Cofalit with the lowest thermal conductivity ( $1.4$  to  $2.1 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  for the Cofalit rock and  $18$  to  $33 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  for aluminium spheres) has shown the better thermal cycle efficiency (79% vs. 71% for discharge efficiency) and the thinner thermocline (26% vs. 33%) than aluminium sphere [43]. The thermal conductivity has to be optimum to ovoid heat increasing between adjacent particles when it is too high and heat discharge rate reduction when it is too low [26]. The different type of filler materials properties useful for DMT storage tank are noted in **Table 1**.

A good filler material of TES system must have thermo physical properties following aspects according to the given requirements of the application fields [50].

**Density:** High density improves energy storage density which reduces the tank volume of the TES system, as consequence reduction of land us. A minimum value of  $2 \text{ M J}\cdot\text{m}^{-3}\cdot\text{K}^{-1}$  is required [51]; **specific heat:** sensible heat storage materials should have high specific heat. High specific heat improves energy



**Figure 5.** Variation of thermocline thickness with discharge time for quartzite rock particle diameters [26].

**Table 1.** Thermal properties and discharge efficiency of some storage materials.

Material properties	$\lambda$ W·m <sup>-1</sup> ·K <sup>-1</sup>	$\rho$ (kg·m <sup>-3</sup> )	$c_p$ J·kg <sup>-1</sup> ·K <sup>-1</sup>	$\rho c_p$ kJ·m <sup>-1</sup> ·K <sup>-1</sup>	$\eta_{dchg}$ (%)	$T_{max}$ (°C)
<b>1. Rocks</b>						
River pebble [44]		2713.6				
Pea pebble [44]	1.2	2506.3				
Sand and quartzite [45]	5.5	2600	973 (200°C)	/	/	
quartzite [26]	5.69	2500	830	2075	90.43	650
Dune sand [35] [46] [47]	1	2600 - 2660	926	/	/	600
River sand	//	//	//	//	//	//
Quartzitic sandstone [42]	5	2733	652 25°C	1783 (25°C)		650
Rhyolite [42]	2.3	2601	785 at 25°C	2045 (25°C)	/	650
Silicate sand	//	//	//	//	//	600
<b>2. Developed materials</b>						
HighTemp concrete [26]	1	2750	780	2519	91.69	≥370
From ash and clay	0.75 - 0.96	1720 - 2140 (R.D)		1700 - 1995	/	1000
From laterite						
Aluminium ceramic [26]	30	3750	916	295	90.03	≥370
S.C ceramic [26]	120	3210	750	2407.5	90.70	
Den stone ceramic pebbles [33]			700	1300		900
Laterite ceramic [48]	1.1 - 2.5		700	1300		900
Industrial waste						
Cast iron [26]	29.3	7900	837	6612.3	89.7	≥370
ACW ceramic [49]	2.1 - 2.4	3120	800 - 1034	2496 - 3226	/	1200

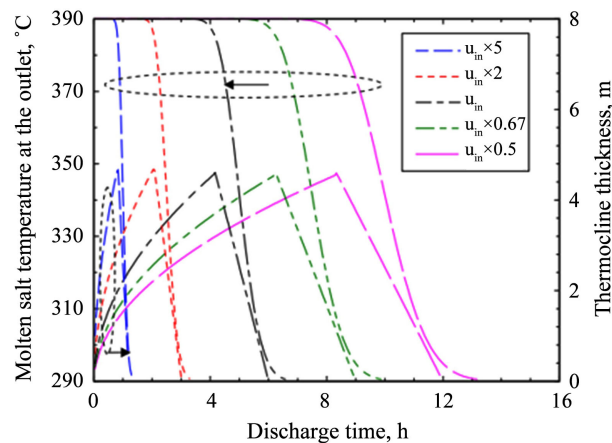
storage density of the system.; **thermal conductivity**: high thermal conductivity increases the thermal charging and discharging rate which is desired. But a maximum value of  $25 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$  is required to avoid thermocline degradation. However, a recent work has showed the low effect of effective thermal conductivity on DMT storage performance [52]; **thermal stability**: the filler materials should not decompose at high temperatures. This gives wider operating temperature range and higher energy storage capacity for the material. Material properties should be stable even after extended thermal cycles of heating and cooling; **chemical stability**: high chemical stability of filler storage materials increases life of energy storage plant making them compatible with the HTF and the storage tank wall; **particle size**: smaller particles enhance thermal stratification, reduce the thickness of the thermocline, improve discharge efficiency thereby the storage efficiency. An optimal value of the particle size could range from 1 cm to 2 cm but should be less than 3 cm [26] [27] [53] depending on the storage material properties and type to reduce heat losses.

### 3.2. Heat Transfer Fluid Effect on Thermocline Storage Effectiveness

Heat transfer fluid must be compatible with the containment materials, storage media and be able to operate in the required temperature range. The HTF defines the maximal temperature to store in the packed bed and the range temperature of the power block (thermodynamic cycle). To have a large temperature range the HTF must be stable at high temperature. The main HTFs are mineral oil, air, molten salt and non-conventional HTFs like vegetal oils and liquid metal. A HTF impacts the thermal energy storage performance through:

**Velocity**: Several studies have shown the impact of the fluid velocity or inlet Reynolds number on thermal performance. Using molten salt as HTF, Xu *et al.* [54] have pointed out the increase of effective discharging time and discharging efficiency at low velocity of the fluid, and shown the possibility to achieve 9.1 h as discharging time at  $3.0085 \times 10^{-4} \text{ m}\cdot\text{s}^{-1}$  of fluid speed with a discharge efficiency of 89.76% and 0.91 h as discharge time at  $6.017 \times 10^{-3} \text{ m}\cdot\text{s}^{-1}$ . But the influence of the velocity is negligible on thermocline development (Figure 6). The similar results have been achieved by Esence *et al.* [36] using monomodal approach and therminol<sup>®</sup> 66 oil as HTF showing also the influence of fluid velocity on thermal stratification. At high velocity the heat flux exchange between the fluid and solid filler becomes lower than the flux conveyed by the fluid, which spreads the thermocline degrading thermal stratification. This degradation may be due to increase of the mixing coefficient [55]. However a possible reduction of heat losses and thermal diffusion can be possible at high fluid velocity, but driving to utilization rate reduction [36] observed experimentally by [34] [56].

**HTF Inlet Temperature Effect**: The inlet temperature or difference between inlet temperature and outlet temperature influences less the thermocline thickness and the discharge efficiency and discharge time [54]. However when the hot

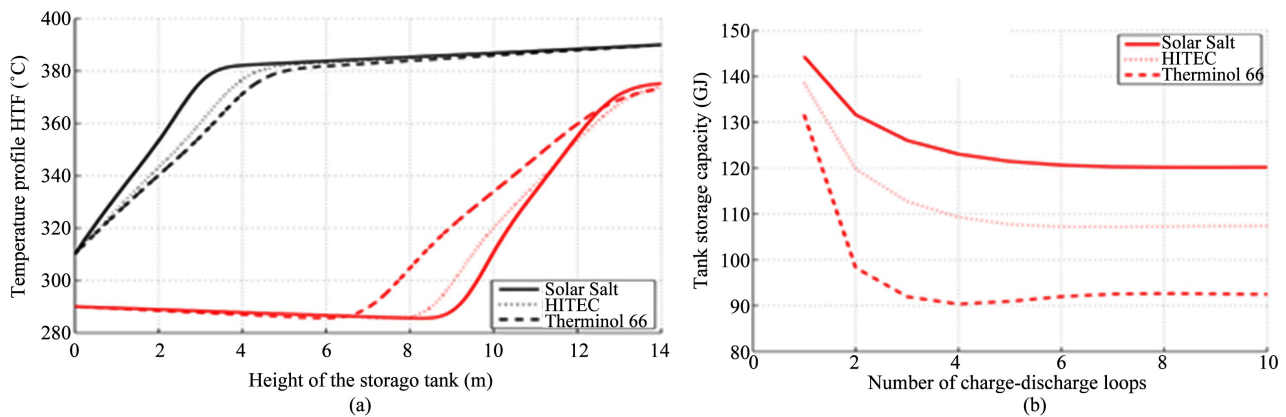


**Figure 6.** Variations in the fluid (molten salt) temperature at the outlet and the thermocline thickness with the discharging time using different fluid inlet velocities [54].

temperature (maximal temperature) of the storage system is too high it can reduce the charge and discharge time, making the storage less effective due to increase of thermal losses at high temperature [57]. In addition when the filler material is encapsulated PCM material, the inlet temperature during charging and discharging affects temperature distribution [58].

**Influence of HTF Properties on Thermal Performance:** The main properties of HTF influencing the performance of thermal energy storage are specific heat, density, kinetic viscosity and thermal conductivity. As these properties differ from one fluid to another, the thickness of the thermocline, the thermal cycle efficiency (charge/discharge) and the stability of tank storage capacity may be also different as shown in **Figure 7**.

The volumetric heat capacity ( $\rho c_p$ ) enables to assess the heat storage density of materials, while the thermal effusivity  $E$  enables to assess their ability to exchange heat. Liquid HTFs like molten salt and oils have good heat capacity and thermal conductivity compared to gasses. This enables liquid/solid systems to operate at low HTF velocity while keeping a good heat transfer coefficient (HTC), which improves thermal stratification [59]. A comparison study between a synthetic oil and rapeseed oil of thermocline performance base on the same sizing condition gave a better thermal performance for the synthetic oil showing thus the depending of thermal storage efficiency on thermal and thermo-physical properties, but rapeseed oil has given a better cost effectiveness [60]. Similar results have been found numerically by Cascetta *et al.* using molten salt, oil and air as HTFs, with a possibility to improve within a large temperature range the performance of the system working with air [61]. The influence of fluid properties like viscosity and thermal conductivity is responsible for maintaining thermocline thickness within the tank. Higher is the viscosity, faster the thermocline moves and may reduce the discharge time thereby the operation time [30]. Moreover, a HTF with very low (limiting the heat transfer between HTF and solid particle) and very high (increasing thermal diffusion) thermal



**Figure 7.** Temperature profile for different HTFs (a), tank storage capacity for different HTFs (b) [57].

conductivity is not suitable. This shows whether the presence of an optimal value of thermal conductivity for better heat transfers between solid and HTF and internal heat diffusion [27]. In addition the storage utilization factor and the exergy efficiency may be better for liquid HTFs (molten salt and oil) than air due to their higher thermo-physical properties [62]. The **Table 2** presents the thermo-physical properties of several fluids for thermal energy storage and heat transfer within a DMT storage tank.

A part from thermo-physical properties a good sensible HTF should have:

**Low melting point:** a low melting point reduce increase the temperature range of the storage system and its stability reducing the waste of energy use for melting; **good chemical stability:** high chemical stability of storage materials increases life of energy storage plant. So the HTF should be compatible with the others elements of the storage system. The indices value and water content should be very low to avoid corrosion of the solid particles, the tank and piping system; **low toxicity:** the HTFs should not be harmful to health of operators and environment. The mineral oil and synthetic oils derive from petroleum are toxic to the environment, that why HTF like vegetal oils have gained considerable interest [63]; **high flammability temperature:** the HTFs should be non-flammable and non-explosive. A high flash point and fire point are required when the HTF is oil; **low vapour pressure:** the HTF should have a low vapor in operational temperature range. High vapor pressure requires pressure withstanding containment at high temperatures. It also requires costly insulation.

### 3.3. Impact of Tank Parameters on Thermal Performance

The tank is the main element of the storage system where there are filler materials. The tank can impact the storage performance through its size (ratio H/D and height), porosity, insulation, position and shape. So these parameters need to be chosen carefully to ensure an optimal operation and to satisfy the required energy. The determination of the tank size is dictated by several operational constraints like the total energy required, the period of the discharge process, the high and low temperature of the HTF, the porosity of the tank and thermo-physical properties of the filler materials and the HTF [60].

**Table 2.** Comparison of thermo-physical properties of some HTFs.

TFS/properties	$\rho$ kg·m <sup>-3</sup>	$c_p$ J·kg <sup>-1</sup> ·K <sup>-1</sup>	$\rho c_p$ kJ·m <sup>-1</sup> ·K <sup>-1</sup>	$\mu$ (Pa s)	$\lambda$ W·m <sup>-1</sup> ·K <sup>-1</sup>	$T_{\min}-T_{\max}$ (°C)
<b>1. Molten salt</b>						
Solar salt [30]	1899	1495	/	0.00326	0.57	221/600
MgCl <sub>2</sub> -KCl [27]	1660	1150	1909	0.005	0.5	426/>500
Na <sub>2</sub> CO <sub>3</sub> -K <sub>2</sub> CO <sub>3</sub> -Li <sub>2</sub> CO <sub>3</sub> [27]	1848	1612	2979	5.9	0.00047	398/>500
ZnCl <sub>2</sub> -NaCl-KCl [27]	1977	900	1779.3	0.0042	0.29	204/>500
HITEC [30] [40]	1640	1560	/	0.00316	0.38	142/538
<b>2. Vegetal oil (NC) (210°C)</b>						
Rapeseed oil [3]	787.7	2492	1963	0.003155	0.143	-21/250
Jatropha curcas oil [3] [63]	778.2	2400 - 2509	1953	0.001781	0.139	3/>200
<b>3. synthetic oil</b>						
<b>Therminol<sup>®</sup> 66oil</b> [64]	885 (200°C)	2195 (200°C)	/	/	0.113 (200°C)	/400
<b>Therminol<sup>®</sup> VP-1</b> [65]	757.7	2464.5		0.000185	0.0865	12/400
sylthermXLT (210°C) [16]	660	2171	1433	0.00023	0.23	-11/260
Coloria [17] [66]	867	2200				
Xceltherm 600 [16]	736	2643	1945	0.00055	0.13	-29/316
<b>4. Liquide Metals (700°C) [27]</b>						
Sodium	798	1256	1002.3	0.00018	57.5	
Lead-bismuth Eutectic	9807	138	1353.4	0.0011	16.8	125/>600
Tin (Sn)	6650	282	1875.3	0.0011	32	232/>6000
Pure lead (Pb)	10196	141	1437.6	0.0014	19.9	327/>600
<b>5. Gasses</b>						
Air (450°C) [64]	0.448/0.50	1080/1190	/	/	0.053	0-infinite

The tank volume is ideal when the solid filler material is not taken into account, and can be calculated as follow:

$$V_{ideal} = \frac{Q_{total}}{\rho_p c_{p,f} (T_h - T_c)} \quad (13)$$

When taking into account the solid filler material, the real volume of the tank should be calculated by:

$$V_{real} = \frac{\rho_p c_{p,f}}{\varepsilon \rho_p c_{p,f} + (1 - \varepsilon) \rho_s c_{ps}} \quad (14)$$

The performance of thermal storage system depending on the effective discharge efficiency, provides the amount of useful energy that the tank can deliver during the discharging process.

### 3.3.1. The Effect of the Ratio H/D and the Height on the Thermocline Effectiveness

The thermal energy storage efficiency increases with the ratio H/D, because high ratio result in higher velocities of the HTFs and consequently a better heat transfer with the solid particles [27] [60] [65]. However when the ratio is higher than around 2.4, the ratio effect becomes negligible [29]. Also the discharge and collection efficiency increase with the height of the vessel [67]. However when the tank is taller than 30 m there is no more improvement and heat loss become more significant [4]. **Figure 8** shows how the ratio H/D and height of the tank can affect the thermocline storage efficiency.

### 3.3.2. Effect of Tank Porosity on Thermal Performance

A low porosity between 0.15 and 0.35 is desirable without compensating much on the effectiveness of the storage system [30].

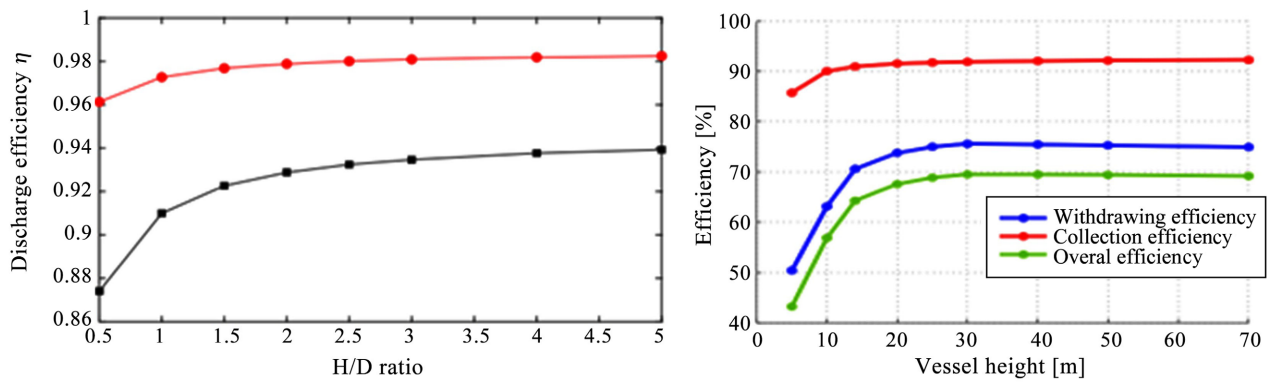
### 3.3.3. The Effective Geometry and Position of Thermocline Storage Tank

Most of the storage tanks studied in the literature are cylindrical [25] [28] [40] [45] [68]. We have also rectangular shape [10] [12] and truncated cone shape [39] [69]. In 2019 based on 2D physical and mathematical model of a novel truncated cone Mao *et al.* have proposed a shell-and-tube TES tank based on enthalpy method. Compare to cylindrical tank the storage efficiency of the truncated cone tank is better and can reach 93% for heat latent storage system [70]. However, concerning the DMT storage, Ortega *et al.* 2017, have showed that, the cylindrical tank has a better thermal performance than truncated cone tank fixing the particle size at 1 cm at the mass flow 0.3 kg/s [53], but in term of mechanical performance the conical geometry is better with worse thermal performance.

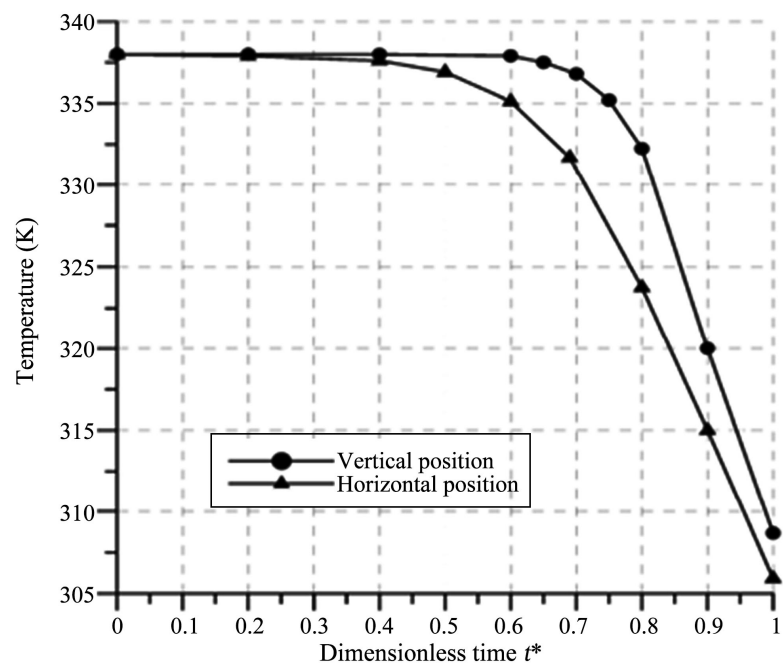
In the literature we did not yet see any study concerning the DMT thermocline tank position on the thermal performance. However a CFD simulation to analyze thermal stratification in domestic hot water storage tank(SMT) have given a better results(stratification and discharge efficiency) for the vertical tank than the horizontal one [68] as presented in **Figure 9**.

### 3.3.4. The Influence of Tank Wall Configuration on Thermal Storage Performance

Based on adiabatic numerical model after tested the influence of heat losses on 1D-2P model, Hoffman *et al.* carried out heat losses through the tank wall for small tanks, but these heat losses may be negligible for large tanks like solar One storage tank [28]. Sometimes the filler materials undergo some amount of volume change due to thermal expansion and mechanical interactions increasing also the tank wall. This expansion of tank wall is from hoop stress due to solid filler material expansion, and may conduct to catastrophic thermal ratcheting and thermocline degradation at high values of hoop stress [9].



**Figure 8.** Effect of the ration H/D on discharge efficiency [27] (left) and the variation of discharge, collection and overall efficiency with the tank height [4] (right).



**Figure 9.** Effect of tank position on thermal energy storage performance.

#### 4. Conclusion

In the short to medium term, packed-bed energy storage is a promising technology to enhance the development of solar thermal energy. Thus several studies have been carried out on thermal storage performance based on modelling giving a good accuracy with experimental data from lab and pilot scale system. This review gave us an overview of the different parameters influencing the thermal storage performance, which are the diameter of the filler materials and their properties, the velocity of the HTF and its properties, the ratio H/D, the height, the geometry and position of the tank, porosity and the inlet geometry in the tank. So for a better sizing of the thermocline storage system, all these parameters should be optimum. A better thermal performance can be achieved when adding an encapsulated phase change materials (PCM) [71] [72].



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## Conflicts of Interest

The authors declare no conflicts of interest regarding the publication of this paper.

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